

SHORT  
COMMUNICATIONS

## Spiro Heterocyclization of Hetareno[*a*]pyrrole-2,3-diones in Reactions with *N*-Alkylanilines

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Received July 12, 2007

DOI: 10.1134/S1070428008090297

Reactions of hetareno[*a*]pyrrole-2,3-diones with *N*-alkylanilines were not reported previously. We have found that 3-aryl-2,4-dihydro-1*H*-pyrrolo[2,1-*c*][1,4]-benzoxazine-1,2,4-triones **Ia** and **Ib** react with *N*-methyl- and *N*-ethylanilines **IIa** and **IIb** at a ratio of 1:1 on heating in boiling toluene (reaction time 0.5–2 h) to give substituted 1-alkyl-3'-aroyl-1'-(2-hydroxyphenyl)-4'-hydroxyspiro[indole-3,2'-pyrrole]-2,5'(1*H*,1'*H*)-diones **IIIa** and **IIIb**.

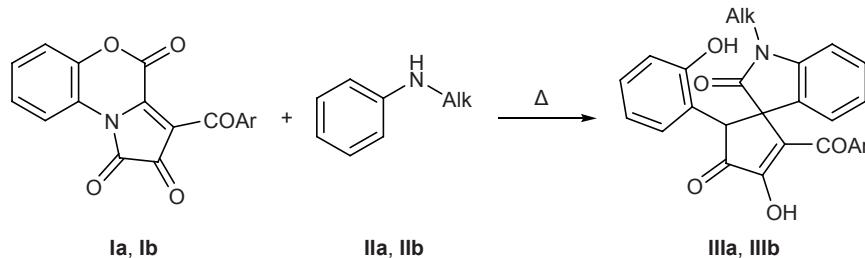
Presumably, the process involves initial electrophilic attack by the C<sup>3a</sup> atom of pyrrolobenzoxazinetrione **I** at the activated *ortho*-position in aniline **II**, followed by closure of indole ring via intramolecular attack by the secondary amino group on the lactone carbonyl carbon atom in the 1,4-oxazine ring and cleavage of the latter at the C<sup>4</sup>–O<sup>5</sup> bond. The transformation sequence is similar to that proposed previously for the reactions of pyrrolobenzoxazinetriones **I** with 3-amino-5,5-dimethylcyclohex-2-en-1-ones, leading to substituted 3'-aroyl-4'-hydroxy-1'-(*o*-hydroxyphenyl)-6,6-dimethyl-6,7-dihydrospiro[indole-3,2'-pyrrole]-2,4,5'(1*H*,1'*H*,5*H*)-triones whose structure was proved by X-ray analysis [1].

The described reaction is a rare example of selective synthesis of difficultly accessible spiro[indole-

3,2'-pyrrole] system containing various substituents in several positions of both heterocyclic fragments.

**3'-Benzoyl-1'-(2-hydroxyphenyl)-4'-hydroxy-1-methylspiro[indole-3,2'-pyrrole]-2,5'(1*H*,1'*H*)-dione (IIIa).** A solution of 0.002 mol of compound **Ia** and 0.002 mol of *N*-methylaniline (**IIa**) in 10 ml of anhydrous toluene was heated for 2 h under reflux (until it became colorless). The mixture was cooled, and the precipitate was filtered off. Yield 77%, mp 268–270°C (from ethyl acetate). IR spectrum,  $\nu$ , cm<sup>−1</sup>: 3244 br (OH), 1707 (C=O), 1621 (PhC=O). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.19 s (3H, CH<sub>3</sub>N), 6.65–7.72 m (13H, H<sub>arom</sub>), 9.59 s (1H, 2"-OH), 12.48 br.s (1H, 4'-OH). <sup>13</sup>C NMR spectrum (DMSO-*d*<sub>6</sub>),  $\delta$ <sub>C</sub>, ppm: 26.78 (CH<sub>3</sub>N), 70.04 (C<sub>spiro</sub>), 108.56–153.72 (C<sub>arom</sub>, C<sup>3'</sup>, C<sup>4'</sup>), 165.28 (C<sup>5'</sup>), 172.92 (C<sup>2</sup>), 188.53 (COPh). Found, %: C 70.40; H 4.27; N 6.54. C<sub>25</sub>H<sub>18</sub>N<sub>2</sub>O<sub>5</sub>. Calculated, %: C 70.42; H 4.25; N 6.57.

**3'-(4-Ethoxybenzoyl)-1-ethyl-1'-(2-hydroxyphenyl)-4'-hydroxyspiro[indole-3,2'-pyrrole]-2,5'(1*H*,1'*H*)-dione (IIIb).** Yield 85%, mp 265–267°C (from ethyl acetate). IR spectrum,  $\nu$ , cm<sup>−1</sup>: 3237 br (OH), 1707 (C=O), 1626 (ArC=O). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.14 t (3H, CH<sub>3</sub>CH<sub>2</sub>N,  $J$  = 7.2 Hz), 1.34 t (3H, CH<sub>3</sub>CH<sub>2</sub>O,  $J$  = 7.1 Hz), 3.70 q and 3.80 q (1H each,



Ar = Ph (**a**), 4-EtOC<sub>6</sub>H<sub>4</sub> (**b**); Alk = Me (**a**), Et (**b**).

$\text{CH}_2\text{N}$ ,  $J = 7.2$  Hz), 4.11 q (2H,  $\text{CH}_2\text{O}$ ,  $J = 7.1$  Hz), 6.64–7.73 m (12H,  $\text{H}_{\text{arom}}$ ), 9.54 s (1H, 2"-OH), 12.24 br.s (1H, 4'-OH).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 11.82 ( $\text{CH}_3\text{CH}_2\text{N}$ ), 14.41 ( $\text{CH}_3\text{CH}_2\text{O}$ ), 56.90 ( $\text{CH}_2\text{N}$ ), 63.40 ( $\text{CH}_2\text{O}$ ), 70.12 ( $\text{C}_{\text{spiro}}$ ), 108.58–153.76 ( $\text{C}_{\text{arom}}$ ,  $\text{C}^3'$ ,  $\text{C}^4'$ ), 162.24 ( $\text{C}^5'$ ), 172.48 ( $\text{C}^2$ ), 187.00 (3'-C=O). Found, %: C 69.38; H 5.06; N 5.71.  $\text{C}_{28}\text{H}_{24}\text{N}_2\text{O}_6$ . Calculated, %: C 69.41; H 4.99; N 5.78.

The IR spectra were measured from samples dispersed in mineral oil on an FMS-1201 spectrometer. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded from solutions in  $\text{DMSO}-d_6$  on a Bruker WP-400 instrument

using tetramethylsilane as internal reference. The purity of the products was checked by TLC on Silufol plates using ethyl acetate as eluent; spots were detected by treatment with iodine vapor.

This study was performed under financial support by the Russian Foundation for Basic Research (project no. 07-03-96036).

#### REFERENCE

1. Racheva, N.L., Shurov, S.N., Aliev, Z.G., and Maslivets, A.N., *Russ. J. Org. Chem.*, 2007, vol. 43, p. 108.